

The Absolute Chirality of (+)-12-Bromo-[2.2]metacyclophane-4-carbonitrile¹

Harald Lehner^a, Helmut Paulus^b, and Karl Schlögl^{a,*}

^a Institut für Organische Chemie, Universität Wien,
A-1090 Wien, Austria

^b Institut für Physikalische Chemie, Technische Hochschule Darmstadt,
D-6100 Darmstadt, Federal Republic of Germany

(Received 8 October 1980. Accepted 21 October 1980)

The absolute configuration of (+)-12-bromo-[2.2]metacyclophane-4-carbonitril [(+)-**4e**] was determined by the *Bijvoet* X-ray diffraction method as (*R*)_p (Fig. 1). This result allows the unambiguous assignment of the absolute chiralities of all optically active 4-monosubstituted, 4,14-homodisubstituted, 4,12- and 4,14-heterodisubstituted [2.2]metacyclophanes chemically correlated with (+)-**4e**. The result is compared with those obtained by other methods for establishing the absolute configuration.

(*Keywords: Bijvoet X-ray diffraction method; Circular dichroism; Exciton model; Kinetic resolution*)

Die absolute Chiralität von (+)-12-Brom-[2.2]metacyclophan-4-carbonitril¹

Die absolute Konfiguration von (+)-12-Brom-[2.2]metacyclophan-4-carbonitril [(+)-**4e**] wurde mit Hilfe der anomalen Röntgenbeugung (*Bijvoet*-Methode) als (*R*)_p (Fig. 1) ermittelt. Damit sind auch die Konfigurationen aller optisch aktiven 4-monosubstituierten, 4,14-homodisubstituierten sowie der 4,12- und 4,14-heterodisubstituierten [2.2]Metacyclophane bekannt, die mit (+)-**4e** eindeutig chemisch korreliert sind. Dieses Ergebnis wird mit jenen verglichen, die mit Hilfe anderer Methoden zur Konfigurationsermittlung erhalten wurden.

Introduction

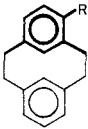
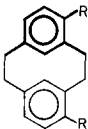
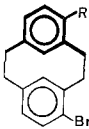
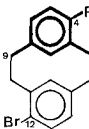
[2.2]Metacyclophanes are of increasing importance as vehicles in stereochemical and conformational studies (cf.²⁻⁵ and references cited therein). In this context we have described the syntheses, optical

resolution and enantiomeric purities of a variety of chiral [2.2]metacyclophanes with various substitution patterns belonging to the point groups C_1 and C_2 , resp.^{1, 2, 6, 7}. A tentative proposal as to the absolute configuration was based on the result of the kinetic resolution of the anhydride of (\pm)-[2.2]metacyclophane-4-carboxylic acid (**1f**) with (–)(*S*)-phenethylamine⁶ and subsequently from the circular dichroism (CD) spectra⁴. These results led to the configuration (*S*)_p for (–)-**1f**.

Kinetic resolutions had been successfully employed to carboxylic acids of systems with related topology, such as ferrocene, benchrotrene, cymantrene, [2.2]paracyclophane, [10]paracyclophane, methano[10]annulene and dihydroethanoanthracene⁸. For the metallocenes and [2.2]paracyclophane the absolute configuration derived from these results had subsequently been confirmed by the X-ray diffraction method⁹.

We report now the absolute chirality of the title compound (+)-**4e** established by the *Bijvoet* method. The knowledge of the configuration of this key compound **4e** renders possible the unambiguous configurational assignment of all optically active [2.2]metacyclophanes which had been previously correlated with (+)-**4e** (see Table 1).

Table 1. Absolute configurations of mono- and disubstituted [2.2]metacyclophanes

							
1	2		3	4			
(–)-a	(–)-a	<i>R</i>	—	—			
(–)-b	(–)-b	CH ₃	—	—			
(+)-c	(+)-c	CH ₂ OH	—	—			
(–)-d	(–)-d	CH ₂ Br	—	—			
(–)-e	(+)-e	CHO	—	—			
(+)-f	(–)-f	COCH ₃	—	—			
(–)-g	(–)-g	COOH	(+)-a	(–)-a			
(+)-h	(+)-h	COO [–]	(+)-b	(–)-b			
(–)-i	(–)-i	COOCH ₃	(+)-c	(–)-c			
(+)-j	(+)-j	CONH ₂	(+)-d	(–)-d			
		CN	(+)-e	(+)-e			

Materials and Methods

Optically pure (+)-**4e** was prepared as described in ref.². The sample was characterized by its mp. (175–177 °C), rotation ($[\alpha]_D^{20} + 9^\circ$ in ethanol) and mass spectrum (m/e 313, 311). The CD-spectrum was recorded at 20 ± 0.1 °C with the

dichrograph Mark III (*Jobin-Yvon*) in ethanol using cylindrical quartz cuvettes of variable path lengths (0.1–1.0 cm). $\Delta\varepsilon(\lambda, \text{nm})$: +23 (202), –21 (216), +6.3 (255), –0.56 (297), cf. Fig. 2.

Crystals for the X-ray analysis were obtained by slow evaporation of an ether solution at room temperature.

The following crystal data were obtained:

monoclinic, space group $P2_1$

$a = 13.635(2)$

$b = 7.899(1)$

$c = 6.345(1) \text{ \AA}$

$\beta = 94.18(1)^\circ$

$D = 1.52 \text{ g} \cdot \text{cm}^{-3}$

($Z = 2$)

4,706 reflections were measured with a STOE Four-circle diffractometer.

Averaging equivalent reflections yielded 2411 symmetry independent reflections, of which 2272 were employed for the refinement. The final weighted R -factor amounted to 2.45%. All crystallographic calculations were performed with the program SHELX*, modified for use at a small computer.

Results and Discussion

The geometry of the skeleton of (+)-**4e** closely resembles that of achiral [2.2]metacyclophanes. Thus the distortions of the single and double bonds (cf. Fig. 1) are very similar to [2.2]metacyclophane^{10,11} itself, its 8,16-dimethyl¹²-, 8,16-dibromomethyl¹³- and 8,16-difluoro-derivative¹⁴ as well as to related tribenzaspirophanes¹¹. As in those compounds the distortions observed in (+)-**4e** are due to a steric compression of the intraannular positions 8 and 16 forcing the benzene rings into a boat like conformation. The rather insignificant differences in the geometric parameters of **4e** as compared with [2.2]metacyclophane are in agreement with earlier findings, showing that the deviations from the C_{2h} -symmetry of the parent hydrocarbon by substitution in the extraannular aromatic positions ought to be very small².

A comparison of the R -factors obtained for the possible configurations** (6.1 and 3.6%, resp.) of **4e** ascribed the configuration as shown in Fig. 1C** to the dextrorotatory enantiomer (+)-**4e**.

Thus, according to previous correlations with other 4,12-heterodisubstituted (**4a–4d**)², with 4,14-heterodisubstituted (**3a–3e**)¹, with 4,14-homodisubstituted (**2a–2j**)¹ and especially with 4-monosubstituted [2.2]metacyclophanes (**1a–1j**)⁶ the absolute configurations as listed in Table 1 result.

This assignment of the absolute configurations of mono- and disubstituted [2.2]metacyclophanes by correlation with the key compound (+)-**4e** is much more straightforward than the recent laborious

* *G. M. Sheldrick*, Cambridge 1976, private communication.

** According to Ref.¹⁵ C-9 (see Table 1) is the "pilot atom" responsible for the orientation of the molecule in applying the (*R*)/(*S*) nomenclature for planar chiral compounds. Accordingly, (+)-**4e** has the configuration (*R*)_p.

deduction of the absolute configuration of 4-methyl-[2.2]metacyclophane (**1a**)¹⁶ from the 1-oxo-derivative¹⁷ and avoids any assumption. The result reported in¹⁶—(*R*)_p for (+)-**1a**—is based on the—albeit

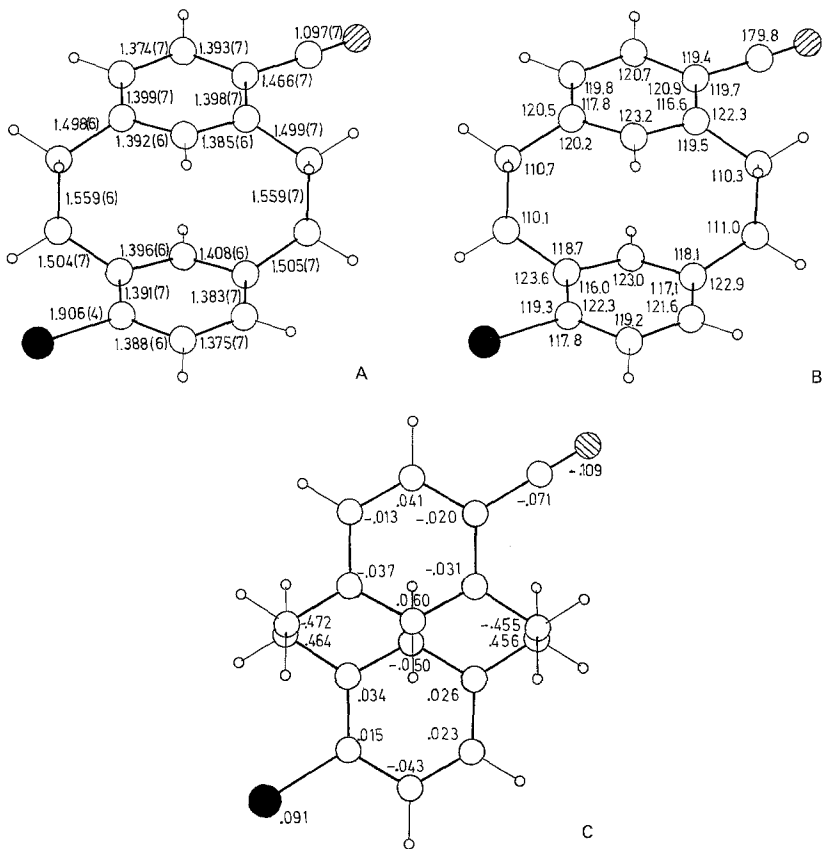


Fig. 1. Structure of (+)-12-bromo-[2.2]metacyclophane-4-carbonitrile (**4e**), ● = Br; ⊙ = N. *A* Bond lengths (Å), *B* Bond angles, *C* Deviations of atom (Å) from the best planes of the benzene rings

very plausible—assumption that methyl has no influence on the sign of the carbonyl *Cotton* effect.

It seemed of interest to compare the configuration derived from the unambiguous X-ray diffraction with the results gained by means of other methods. The method of kinetic resolution⁶ obviously failed in the case of the [2.2]metacyclophane system. As mentioned above this method had given good results in several other cases, especially for [2.2]paracyclophane-carboxylic acid⁹.

At the present it is not clear whether the failure of this method is due only to the special topology of the [2.2]metacyclophane system where the difference between "above" and "beneath" for a substrate approaching the 4-position on which this method is based, is not so pronounced as for instance in [2.2]paracyclophane. It should be emphasized in this context, however, that only the *relative* configurations of the metacyclophanes were relevant for the studies as described in refs.^{1,2,4,6}. The conclusions drawn are *independent* from the *absolute* configuration.

A further method for the assignment of the absolute configuration of bichromophoric systems is based on the application of the coupled oscillator model¹⁸. However, in analogy to the 4-monosubstituted [2.2]metacyclophanes⁴ studied so far (**1a-1j**) no bisignate *Cotton* effect is observed either in the 1L_b - nor in the 1L_a -transition region of the CD spectrum of (+)-**4e** (Fig. 2), indicating that the electronic interactions

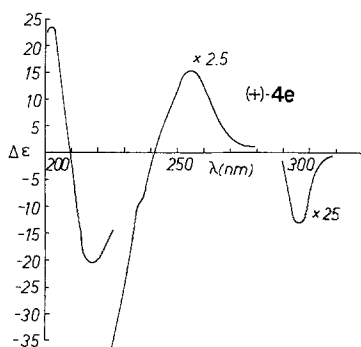


Fig. 2. CD-spectrum of (+)-**4e** in ethanol

between the two aromatic moieties in **4e** are weak. Accordingly, a band splitting of the 1L_a -*Cotton* effect will occur only in the CD spectra of derivatives where both aromatic rings carry substituents with large spectroscopic moments⁴. Thus—and since the molecule exhibits C_1 -symmetry—the point dipole approximation cannot be applied to (+)-**4e**. For this purpose, a derivative like (+)-**2h** with two methoxy-carbonyl ligands was chosen.

The coordinate system was chosen so that the x -axis transfixes the centres of both aromatic rings and the z -axis is parallel to the C_2 -axis of the molecule. The 1L_a -transition moment of (R)_p-**2h** comprises the angles $r = 106^\circ$, $t = 101^\circ$ and $v = 160^\circ$ with the coordinate axes, so that according to equations (4) and (5) in ref.⁴ $\nu_A - \nu_B > 0$ results for the

band splitting and $R_A > 0$, $R_B < 0$ for the rotational strengths of the bands with A and B symmetry*.

A comparison with the actual CD spectrum yields the configuration $(R)_p$ for $(-)\text{-2h}$. On the basis of the chemical correlation this is in agreement with the X-ray configuration determination for $(+)\text{-4e}$ (cf. Table 1).

Acknowledgements

Generous supports by the "Fonds zur Förderung der wissenschaftlichen Forschung Österreich" (Project 3033) and by the "Hochschuljubiläumstiftung der Stadt Wien" are gratefully acknowledged.

References

- 1 Stereochemistry of Planar Chiral Compounds, Part 8. Part 7: *Krieger, Ch., Lehner, H., Schlögl, K.*, Mh. Chem. **107**, 195 (1976).
- 2 *Keller, H., Krieger, Ch., Langer, E., Lehner, H., Derflinger, G.*, Liebigs Ann. Chem. **1977**, 1296.
- 3 *Keller, H., Lehner, H.*, Liebigs Ann. Chem. **1978**, 595.
- 4 *Keller, H., Krieger, Ch., Langer, E., Lehner, H., Derflinger, G.*, Tetrahedron **34**, 871 (1978).
- 5 *Krois, D., Langer, E., Lehner, H.*, Tetrahedron **36**, 1345 (1980).
- 6 *Kainradl, B., Langer, E., Lehner, H., Schlögl, K.*, Liebigs Ann. Chem. **766**, 16 (1972).
- 7 *Glitzmann, C., Langer, E., Lehner, H., Schlögl, K.*, Mh. Chem. **106**, 763 (1975).
- 8 For a review see: *Schlögl, K.*, in *Methodicum Chimicum*, Vol. 1, Part 1 (*Korte, F.*, ed.), p. 235. Stuttgart: G. Thieme. 1973; p. 227. New York: Academic Press. 1974.
- 9 See *Klyne, W., Buckingham, W.*, Atlas of Stereochemistry, 2nd., Vol. 1 and 2. London: Chapman and Hall. 1977.
- 10 *Brown, C. J.*, J. Chem. Soc. (London) **1953**, 3278.
- 11 *Kai, Y., Yasuoka, N., Kasai, N.*, Acta Cryst. **B 33**, 754 (1977); see also *Schuster, J. C., Völlenkle, H.*, Mh. Chem. **109**, 1155 (1978).
- 12 *Hanson, A. W.*, Acta Cryst. **15**, 956 (1962).
- 13 *Mathew, M.*, Acta Cryst. **B 24**, 530 (1968).
- 14 *Hanson, A. W.*, Acta Cryst. **B 31**, 2352 (1975).
- 15 *Cahn, R. S., Ingold, C. K., Prelog, V.*, Angew. Chem. **78**, 413 (1966); Intern. Ed. **5**, 385 (1966).
- 16 *Nakazaki, M., Hirose, Y., Shimizu, T., Suzuki, T., Ishii, A., Makimura, M.*, J. Org. Chem. **45**, 1428 (1980).
- 17 *Mislow, K., Brzechffa, M., Gschwend, H. W., Puckett, R. T.*, J. Amer. Chem. Soc. **95**, 621 (1973).
- 18 See e.g. *Haas, G., Hulbert, P. B., Klyne, W., Prelog, V., Snatzke, G.*, Helv. Chim. Acta **54**, 491 (1971).

* The sign of $\nu_A - \nu_B$ has erroneously been interchanged in ref.⁴ resulting in the antipodal configuration for $(+)\text{-2h}$.